

Nano sized *Moringa oleifera* an Effective Strategy

For Pb(II) ions Removal from Aqueous Solution

Laila H. Abdel-Rahman^a, Ahmed M. Abu-Dief^{a*}, M. A. Abd- El Sayed^b

And Mallak Megalea Zikry^b

^a*Chemistry Department, Faculty of Science, Sohag University, 82534 Sohag, Egypt*

^b*Medicinal and Aromatic Plants Researches Department, Horticulture Research Institute (HRI), Agri. Res. Center (ARC), Giza, Egypt*

*Corresponding author. Tel.: +0201098856153; fax: +020934601159

E-mail address: ahmed_benzoic@yahoo.com (A. M. Abu-Dief)

Table S1: Pb²⁺ ions Removal Efficiency and q_e at Different Initial Concentration

C _o (mg/L)	C _e (mg/L)	Pb ²⁺ ions R. E %	q _e
10	0.42	95.83	1.6
20	1.13	94.36	3.15
50	4.16	91.68	7.64
100	14.58	85.42	14.24
200	49.32	75.34	25.11
300	93.24	68.92	34.46
400	172.56	56.86	37.91

Table S2: Pb²⁺ ions removal efficiency q_e at initial concentration of 20 mg/L at different pH values.

pH	C _o (mg/L)	C _e (mg/L)	Pb ²⁺ R. E %	q _e
2	20	7.72	61.4	2.05
4	20	3.65	81.76	2.73
5	20	1.13	94.36	3.15
6	20	1.92	90.4	3.01
7	20	2.1	89.5	2.98
8	20	3.27	83.66	2.79
10	20	3.14	84.28	2.81

Table S3: Pb²⁺ ions removal efficiency and q_e at initial concentrations 20 mg/L and different biosorbent dosage

Biosorbent dosage	C _o (mg/L)	C _e (mg/L)	Pb ²⁺ ions R. E %	q _e
25	20	6.75	66.25	26.5
50	20	5.9	70.5	14.1
100	20	4.14	79.3	7.93
200	20	1.87	90.67	4.54
300	20	1.13	94.36	3.15
400	20	0.96	95.2	2.38
500	20	0.73	96.36	1.93

Table S4: Effect of contact time on Pb²⁺ ions removal efficiency and q_e at different Initial concentrations (10, 30, 50) mg/L by *Moringa oleifera*

Time (min)	Pb R. E % at C _o (10 mg/L)	Pb R. E % at C _o (30 mg/L)	Pb R. E% at C _o (50 mg/L)	q _t at C _o (10mg/L)	q _t at C _o (30mg/L)	q _t at C _o (50mg/L)	C _t at C _o (10mg/L)	C _t at C _o (30mg/L)	C _t at C _o (50mg/L)
20	78.55	73.01	69.3	1.4	4.1	6.69	1.61	5.41	9.84
40	86.3	80.88	80.79	1.52	4.3	7.20	0.9	4.1	6.74
60	95.83	90.76	91.68	1.57	4.53	7.52	0.57	2.81	4.9
80	96.15	91.19	92.2	1.58	4.59	7.54	0.52	2.5	4.75
120	96.21	91.4	92.5	1.58	4.6	7.59	0.503	2.4	4.65

Table S5: Effect of temperature on Pb²⁺ ions removal efficiency and q_e at different initial concentrations (10, 20, 50) mg/L by *Moringa oleifera*

Temp. (°C)	Pb R. E % at C _o (10 mg/L)	Pb R. E% at C _o (20 mg/L)	Pb R. E % at C _o (50 mg/L)	q _e at C _o (10mg/L)	q _e at C _o (20mg/L)	q _e at C _o (50mg/L)	C _e at C _o (10mg/L)	C _e at C _o (20mg/L)	C _e at C _o (50mg/L)
25	95.16	94.36	91.68	1.586	3.15	7.64	0.48	1.13	4.16
30	95.83	94.7	92.2	1.59	3.16	7.68	0.42	1.06	3.9
40	96.7	95.5	93.2	1.61	3.18	7.77	0.33	0.9	3.4
50	96.2	95.3	92.76	1.6	3.178	7.73	0.38	0.93	3.62

Table S6: Isotherm constants of Pb²⁺ ions biosorption on *Moringa oleifera* at various temperatures.

T (K)	Langmuir		Freundlich			
	q _m	b	R ²	n	Kf	R ²
298	15.65	0.23	0.9974	1.38	2.76	0.998
303	14.78	0.27	0.9843	1.43	2.98	0.9996
313	13.94	0.36	0.9748	1.484	3.4	1
323	14.34	0.32	0.9946	1.483	3.41	1

Table S7: A dimensionless constant separator factor (R_L) for Langmuir type biosorption process.

C_o (mg/L)	R_L at 25°C	R_L at 30°C	R_L at 40°C	R_L at 50°C
10	0.303	0.27	0.217	0.238
20	0.178	0.156	0.122	0.135
50	0.08	0.069	0.053	0.059