# An Efficient One-Pot Three-Component Synthesis of Some New Polyhydroquinolines via Enaminone Intermediates 

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For a wide spectrum of pharmacological effects of polyhydroquinolines, this study introduces a developed safe, simple, higher yields and fast method for the synthesis of some new hexahydroquinoline derivatives using one-pot three-component cyclocondensation reaction, via the reaction of 1,3-cyclohexanedione with primary amine and arylidinemalononitrile or salicylaldehyde derivatives. The prepared compounds were reacted with different reagents as $N, N$-dimethylformamide dimethylacetal, acetic anhydride, sulphuric acid, and hydrazine hydrate forming several polycyclic hexahydroquinoline and acridine derivatives. All these new compounds have been characterized by spectral data and expected to be effective pharmaceutical drugs.
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## INTRODUCTION

Quinoline and some of its derivatives are the most widespread $N$-heterocyclic compounds associated into the structures of most pharmaceutical and antimicrobial drugs [1]. Recently, much interest has been dedicated to the synthesis of polyhydroquinoline compounds because of their diverse therapeutic and pharmacological properties. Various quinoline derivatives characterize moderate toxicity and central nervous system stimulants [2,3]. Some fluoroquinolines, for example, 8-difluoro-methoxy-1-ethyl-6-fluoro-1,4-dihydro-7-[4-(2-methoxyphenyl)-1-pipernzinyl]-4-oxoquinoline-3-carboxylic acid] and 7-(3,4-dihydro-4-phe nyl-piperidnyl)-1,4-dihydro-6-fluoro-1-methyl-8-trifluorome thyl-4-oxo-quinolone-3-carboxylic acid have been reported as inhibitors for human immunodeficiency virus-1 replication through interference with the transcription process [4-6]. The known quinoline nucleus existed in different natural compounds like quinones, chloroquine, and papaverine $[7,8]$. The two former nuclei are antimalarial natural products, whereas the third one is a smooth muscle relaxant, a coronary vasodilator, an antispasmodic drug, and also primarily used in the treatment of visceral spasm and vasospasm (especially those involving the heart and the brain) $[9,10]$. Additionally, naturally occurring quinolone derivatives as 2-methyl-1,2,3,4-tetra-hydroquinoline exist in human brain. Dynemycin is acting as antitumor and antibiotic effectives
[11]. Different 1,2,3,4-tetrahydroquinolines have been tested as potential drugs, for example, schistosomicide, nicainoprol [12], and oxamniquine [13] are antiarrhythmic drug, and virantmycin [14] is an antibiotic drug. Tetrahydroquinoline L-689,560 is one of the most useful NMDA antagonists yet found $[15,16]$ (Fig. 1). In continuation to our previous work [17], this study has been undertaken to synthesize some new hexahydroquinolines via one-pot three-component cyclocondensation reaction of 1,3-cyclohexanedione, aryledenmalononitrile, and amine as reagents in aqueous medium expected to have an effective pharmaceutical activities (Scheme 1), the higher effective of the quinolone and acridine encouraged to synthesis of 16 new compounds of these nuclei.

## RESULTS AND DISCUSSION

Firstly, the protocol for the previously mentioned reaction has been established without any catalyst in refluxing aqueous condition, the desired product not prepared. The same reaction was employed incorporating a small amount of triethylamine as catalytic amount under the same reflux condition, so an excellent yield (about $80 \%$ ) of the desired hexahydroquinolines has been obtained. The target compounds 2-amino-3-cyanohexahydroquinoline derivatives 2a-e were prepared in a very good yield, via a one-pot three-component


Figure 1. The oxamniquine, nicainoprol, virantmycin, and tetrahydroquinoline L-689,560.

Scheme 1. Synthesis of hexahydroquinoline 3a,b and octahydroacridine derivatives $\mathbf{2 e - f}$.

cyclocondensation reaction of 1,3-cyclohexanedione $\mathbf{1}$ with $p$-toluidine and a series of substituted arylidinemalononitrile (benzylidinemalononitrile, $p$-chloro benzylidinemalononitrile [17], and $p$-methoxybenzylidine malono-nitrile) forming 2a-c derivatives, in addition to the reaction of 1,3 -cyclohexanedione $\mathbf{1}$ with diamens examply (1,2-diaminoethane or 1,4-diaminobenzene) and $p$-chlorobenzylidinemalono-nitrile in refluxing ethanol containing a catalytic amount of triethylamine. The condensation reaction of 1,2-diaminoethane took place via the two amino groups but of 1,4-di-aminobenzene took place via one amino group only forming ethane-1,2-bis-[2-amino-1-(4-methylphenyl)-5-oxo-4-(4-chlorophenyl)(1,4,5,6, 7,8-hexahydroquinolone-3-carbonitrile] and 2-amino-4-(4chloro phenyl)-1-(4-aminophenyl)-5-oxo-1,4,5,6,7,8-hexahy droquinoline-3-carbonitrile derivatives $\mathbf{2 d}, \mathbf{e}$, respectively (Scheme 1). The spectral analyses of compound 2 e pointed to the presence of an ethanol molecule joined with octahydroacridine moiety during the crystal lattice building. The molecular structures of the products 2a-e were illustrated from their IR, ${ }^{1} \mathrm{H}-\mathrm{NMR}$, and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectral analyses. The empirical mechanism for this transformation can be rationalized through two different steps [18]. Step 1 involves the formation of the required cyclohex-2enaminones from the reaction of 1,3-cyclohexanedione with
the amines under aqueous conditions. The formed enaminone contains a nucleophilic character of both of enamine and enone [19]. Step 2 comprises Michael addition reaction with intramolecular cyclization between enaminone and different arylidenemalononitriles affording the final product hexahydroquinoline derivatives $\mathbf{2 a - e}$ (Scheme 2).

Moreover, under the same previous condition, two molar ratio 1,3-cyclohexandione $\mathbf{1}$ was reacted with equimolar ratio of the aldehyde, namely, salicylaldehyde or $p$-bro mosalicylaldehyde and ethyl glycinate hydrochloride in presence of two molar ratio of triethylamine in refluxing ethanol to afford ethyl-2-[9-(2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-decahydroacridin-10-yl]acetate and ethyl-2-[9-(3-bromo-2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6, 7,8,9,10-deca-hydroacridin-10-yl]-acetate [20] 3a,b, respectively, in good yields $80 \%$ (Scheme 1). In our previous work, compound 3b was obtained with another synthetic route [20] that leads to relatively lower yield. The structures of these compounds were established using IR, ${ }^{1} \mathrm{H}-\mathrm{NMR}$, ${ }^{13} \mathrm{C}-\mathrm{NMR}$, and MS spectral analyses.

When an equimolar ratio of compound $\mathbf{2 b}$ was subjected to react with $N, N$-dimethylformamide dimethylacetal in boiling dimethylformamide (DMF) resulting in the formation of [9-(2-hydroxyphenyl)- $N^{\prime}$-[3-cyano-1-(4-meth ylphenyl)-5-oxo-1,4,5,6,7,8-hexahydroquinolin-2-yl]- $N, N$ dimethylimidoformamide (4). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum (DMSO$d_{6}$ ) revealed a singlet signal at $\delta 7.67$ due to the olefinic proton $\mathrm{CH}=\mathrm{N}$ and two singlet signals due to the dimethylamino group at $\delta 2.90$ and 2.34 ppm (cf. experimental).

However, the reaction of the target compound $\mathbf{2 b}$ with refluxing acetic anhydride leads to the formation of pyrimidohexahydroquinoline derivative 5 in $68 \%$ yield. During an attempt to recrystallize compound 5 from DMF, a DMF molecule was joined with pyrimidohexahydroquinoline moiety during the crystal lattice building. The presence of the DMF molecule was proven in the basis of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}$-NMR spectral analyses. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right)$ spectrum pointed to a

Scheme 2. The reaction mechanism of formation of a hexahydroquinoline derivatives (2a-e).




singlet signal at $\delta 7.97$ due to CH group and two singlet signals at $\delta 2.90$ and 2.75 due to two $\mathrm{CH}_{3}$ groups of DMF molecule. Also, ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO- $d_{6}$ ) spectrum showed three signals of $\mathrm{C}=\mathrm{O}$ carbon at 196.09 of DMF , in addition to another two carbonyl carbons at 172.70 and 162.13 ppm of pyrimidohexahydroquinoline moiety and two $\mathrm{CH}_{3}$ carbons at 21.45 and 21.27 of DMF in addition to another $\mathrm{CH}_{3}$ carbon at 28.36 ppm of pyrimidohexahydroquinoline moiety. The reaction of compound $\mathbf{2 b}$ with concd sulphuric acid took place at $25^{\circ} \mathrm{C}$ via the acid hydrolysis of the cyano group and the oxidation reaction of the amino group to form the keto carboxylic acid derivative 6. The structural elucidation of the synthesized compounds 5 and 6 was carried out by IR, ${ }^{1} \mathrm{H}-\mathrm{NMR}$, and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectral analyses, where IR spectrum of compound 6 revealed the existence of the three strong stretching vibrations at $v 1705,1671$, and $1631 \mathrm{~cm}^{-1}$ due to three carbonyl groups functions of $2,5-$ dioxo-octahydroquinoline-3-carboxylic acid. The keto acid derivative 6 was allowed to react with absolute ethanol in the presence of concd sulphuric acid under reflux for 4 h to form the expected corresponding ethyl
ester derivative 7. The structure of this compound has been further confirmed by refluxing the parent compound 2-amino-3-cyano-hexahydroquinoline derivative $\mathbf{2 b}$ with sulphuric acid in absolute ethanol for 4 h . The spectral data of the product were typical with those of compound 7 (Scheme 3). The condensation reaction of compound 7 with equimolar amounts of phenyl hydrazine and catalytic amount of trifloroacetic acid was studied. This reaction took place only at the carbonyl group of cyclohexenone moiety resulting in phenylhydrazone derivative $\mathbf{8}$ in a good yield (Scheme 3). The structures of these new products 7 and $\mathbf{8}$ have been unambiguously confirmed by IR, ${ }^{1} \mathrm{H}-\mathrm{NMR}$, and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectral analyses.

Basic hydrolysis of compound 3b affording the corresponding carboxylic acid derivative [9-(3-bromo-2-hydroxyphenyl)-2, 3, 4, 5, 6, 7, 8, 9-octahydroacridin$10(1 H)-\mathrm{yl}]$ acetic acid (9), during its recrystallization from ethanol, an ethanol molecule was joined with octahydroacridine moiety during the crystal lattice building. IR, ${ }^{1} \mathrm{H}-\mathrm{NMR}$, and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ analyses gave a good evidence for its structure. The hydrazide derivatives

Scheme 3. The reaction take place on 2-amino-4-(4-chlorophenyl)-1-(4-methylphenyl)-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carbonitrile (2b).


Scheme 4. The reaction take place on ethyl-2-[9-(3-subsituted-2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-decahydro-acridin-10-yl] acetate (3a,b).

$\mathbf{1 0 a}, \mathbf{b}$ were prepared via the reaction of the acridine derivatives 3a,b with hydrazine hydrate in boiling ethanol. When compound 10a was reacted with $p$ chlorobenzaldehyde in DMF via a nucleophilic condensation reaction, [9-(2-hydroxyphenyl) $]-N^{\prime}-[(1 E)-$ ethylidene]-2-[2,3,4,5, 6, 7,8,9-octahydroacridin-10(1H)yl]acetohydrazide (11) has been formed in $50 \%$ yield (Scheme 4). All the new compounds were investigated with spectral analyses as shown in the experimental section.

## CONCLUSION

Regarding to the importance of the polyhydroquinolines effects, a developed safe method for the synthesis of some new hexahydroquinoline derivatives using one-pot three-component cyclocondensation reaction, via the combination of 1,3-cyclohexanedione with primary amine and arylidinemalononitrile or the combination of 1,3-cyclohexanedione with primary amine and salicylaldehyde derivatives was carried out.

## EXPERIMENTAL

All chemical reagents and instruments were bought from Aldrich, Merck, and used directly without further purification. Our products were investigated by IR spectra measured on a Nicolet 710 Fourier transform infrared spectroscopy spectrometer. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra were recorded in deuterated dimethyl sulfoxide at 400 and 100 MHz on a BRUKER spectrometer and DELTA 2-NMR spectrometer using tetramethylsilane as an internal reference. The purity of the substances and the progress of the reactions were screened on thin-layer
chromatography, and all melting points are uncorrected and were recorded on Melt-Tem II melting point apparatus.

Synthesis of 2-amino-1-(4-methylphenyl)-5-oxo-4-(aryl)-1,4,5,6,7,8-hexahydroquinoline-3-carbonitrile (2a-c) (general procedure). To a solution of 1,3-cyclohexanedione $(3.36 \mathrm{~g}, 0.03 \mathrm{~mol})$ and $p$-toludine $(3.21 \mathrm{~g}, 0.03 \mathrm{~mol})$ in ethanol ( 40 mL ), a catalytic amount of triethylamine was added. The reaction mixture was heated under reflux for 3 h. Arylidinemalononitrile, namely, benzylidenemalono trile, 4-chlorobenzylidenemalononitrile, and 4-methoxy benzylidenemalononitrile ( 0.03 mol ) was added to the reaction mixture while reflux for another 3 h . The separated solid was filtered on hot washed with ethanol and crystallized from the suitable solvent.

2-Amino-4-phenyl-1-(4-methylphenyl)-5-oxo-1,4,5,6,7,8-he xahydroquinoline-3-carbonitrile (2a). Crystalized as brown crystal from ethanol; yield $70 \%$; mp $290-292^{\circ} \mathrm{C}$; IR $\left(\lambda_{\max }, \mathrm{cm}^{-1}\right): 3470 \mathrm{~m}, 3312 \mathrm{~m}\left(\mathrm{NH}_{2}\right), 3062-3030 \mathrm{w}$ $\left(\mathrm{CH}_{\text {arom. }}\right), 2946-2888 \mathrm{~m}\left(\mathrm{CH}_{\text {aliph. }}\right), 2178 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1632$ s $\quad(\mathrm{C}=\mathrm{O}), \quad 1610 \mathrm{~s} \quad(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \quad(400 \mathrm{MHz}$, DMSO- $d_{6}$ ), $\delta$ ppm: $7.31-7.16$ (m, $9 \mathrm{H}, \mathrm{CH}_{\text {arom. }}$ ), 5.26 ( s , $2 \mathrm{H}, \mathrm{NH}_{2}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.51(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar})$, 2.40 (s, $3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{Ar}$ ), 2.29-2.19 (t, $2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 1.94-1.81 (t, 2H, CH2-C=C), 1.66-1.61 (m, 2H, CH $2_{2}$ $\left.\mathrm{CH}_{2}-\mathrm{CH}_{2}\right)$; ${ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}$ : 195.47 ( $\mathrm{C}=\mathrm{O}$ ), $152.91,151.59,147.08,139.84,134.11$ (5-C), 131.04, 130.02, 128.82, 127.19, 126.66 (9-CH), 121.88 (C), $113.18(\mathrm{C} \equiv \mathrm{N}), 61.03(\mathrm{C}), 56.53\left(\mathrm{CH}_{2}-\right.$ $\mathrm{C}=\mathrm{O}), 36.53(\mathrm{CH}), 28.20,21.19\left(2-\mathrm{CH}_{2}\right), 21.10\left(\overline{\mathrm{CH}_{3}}\right)$; analysis: calculated for $\mathrm{C}_{23} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}$ (355.43): C, 77.71; H, 5.96; N, 11.82\%. Found: C, 77.65; H, 5.81; N, 11.97\%.

2-Amino-4-(4-chlorophenyl)-1-(4-methylphenyl)-5-oxo-1,4, 5,6,7,8-hexahydroquinoline-3-carbonitrile (2b). Crystalized as pale yellow crystal from DMF; yield: $79 \%$; mp $300^{\circ} \mathrm{C}$; IR $\left(\lambda_{\max }, \mathrm{cm}^{-1}\right): 3472 \mathrm{~s}, 3325 \mathrm{~s}\left(\mathrm{NH}_{2}\right), 3032 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right)$, 2972-2879 w ( $\mathrm{CH}_{\text {aliph. }}$ ), $2177 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1631 \mathrm{~s}(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 400 MHz, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 7.39-7.29$ (m, $\left.8 \mathrm{H}, \mathrm{CH}_{\text {arom. }}\right), 5.32\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right)$, $4.51(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 2.4\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.23-2.19$ ( $\mathrm{t}, J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), $1.93-1.81\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}-\right.$ $\mathrm{C}=\mathrm{C}), 1.65-1.61\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 195.45(\mathrm{C}=\mathrm{O}), 153.14$, 151.70, 146.07, 139.86, 133.96, 131.26 (6-C), 131.03, 130.10, 129.13, 128.79 ( $8-\mathrm{CH}$ ), 121.78 (C), 112.72 (C三N), $60.26(\mathrm{C}), 36.45(\mathrm{CH}), 36.21,28.23,21.22$ (3$\left.\mathrm{CH}_{2}\right), 21.06\left(\mathrm{CH}_{3}\right)$; analysis: calculated for $\mathrm{C}_{23} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{OCl}$ (389.87): C, 70.85 ; H, 5.18 ; N, 10.78\%. Found: C, 70.76; H, 5.27; N, 10.71\%.

2-Amino-4-(4-methoxyphenyl)-1-(4-methylphenyl)-5-oxo-1, 4,5,6,7,8-hexahydroquinoline-3-carbonitrile (2c). Crystalized as brown crystal ethanol; yield $72 \%$; mp $252^{\circ} \mathrm{C}$; IR ( $\lambda_{\text {max }}$, $\left.\mathrm{cm}^{-1}\right): 3468 \mathrm{~m}, 3318 \mathrm{~m}\left(\mathrm{NH}_{2}\right), 3039 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right)$, $2941-$ 2838 w ( $\mathrm{CH}_{\text {aliph. }}$ ), $2180 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1634 \mathrm{~s}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-$ NMR ( $400 \mathrm{MHz}, ~ D M S O-d_{6}$ ), $\delta \mathrm{ppm}$ : 7.38-6.88 (m, 8H,
$\mathrm{CH}_{\text {arom. }}$ ), $5.12\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.48$ ( s , $1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 3.75\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{O}-\mathrm{CH}_{3}\right), 2.41\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{Ar}\right)$, 2.21-2.19 ( $\left.\mathrm{t}, J=4 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 1.94-1.81(\mathrm{t}, 2 \mathrm{H}$, $\mathrm{CH}_{2}-\mathrm{C}=\mathrm{C}$ ), $1.66-1.61\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( 100 MHz, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 195.36$ ( $\mathrm{C}=\mathrm{O}$ ), 158.39, 152.43, 151.47, 139.81, 139.41, 134.25 (6-C), 131.02, $130.02,128.23,114.37$ ( $8-\mathrm{CH}$ ), 121.82 (C), 113.67 (C三N), $61.59(\mathrm{C}), 55.59\left(\mathrm{O}-\mathrm{CH}_{3}\right), 36.59\left(\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 35.72(\mathrm{CH})$, 28.17, $21.21\left(2-\mathrm{CH}_{2}\right), 21.15\left(\mathrm{CH}_{3}\right)$; analysis: calculated for $\mathrm{C}_{24} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{2}$ (385.45): $\mathrm{C}, 74.78 ; \mathrm{H}, 6.02 ; \mathrm{N}, 10.90 \%$. Found: C, 74.69 ; H, 6.15 ; N, 10.83\%.

Synthesis of ethane-1,1'-1,2-diyl-bis-[2-amino-5-oxo-4-(4-chlorophenyl)-1,4,5,6,7,8-hexahydroquinolone-3-carbonitrile] and 2-amino-1-(4-aminophenyl)-4-(4-chlorophenyl)-5-oxo-1,4, 5,6,7,8-hexahydroquinoline-3-carbonitrile (2d,e). To a solution of 1,3 -cyclohexanedione ( $0.896 \mathrm{~g}, 0.008 \mathrm{~mol}$ ) and 1,2-diaminoethane or 1,4-diaminobenzene ( 0.004 mol ) in ethanol ( 20 mL ), a catalytic amount of triethylamine was added, and the reaction mixture was heated under reflux for 3 h . 4-Chlorobenzylidene- malononitrile ( 1.52 g , 0.008 mol ) was added to the reaction mixture while reflux for additionally 5 h . The reaction mixture was concentrated and cooled; the formed precipitate was filtered off and crystallized from the proper solvent.

Ethane-1,1'-1,2-diyl-bis-[2-amino-1-(4-methylphenyl)-5-oxo-4-(4-chlorophenyl)(1,4,5,6,7,8-hexahydroquinolone-3-carbonitrile] (2d). Brown crystal from ethanol; yield 73\%; mp 210-212 C; IR $\left(\lambda_{\max }, \mathrm{cm}^{-1}\right): 3313 \mathrm{~m}, 3201 \mathrm{~m}\left(\mathrm{NH}_{2}\right), 3051 \mathrm{w}$ $\left(\mathrm{CH}_{\text {arom. }}\right), 2950-2872 \mathrm{w}\left(\mathrm{CH}_{\text {aliph. }}\right), 2176 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1650 \mathrm{~s}$ $(\mathrm{C}=\mathrm{O}), 1613 \mathrm{~s}(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right)$, Sppm: 7.33-7.18 (m, 8H, CH arom.), 6.13 (s, 4H, $2 \mathrm{NH}_{2}$, disappeared by $\mathrm{D}_{2} \mathrm{O}$ ), $4.44(\mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{CH}-\mathrm{Ar}), 3.97-3.94(\mathrm{t}$, $\left.J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}\right), 3.74-3.71\left(\mathrm{t}, J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}-\right.$ $\mathrm{N})$, 2.86-2.82, $2.56-2.51$ ( $\mathrm{t}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 2.27-2.24 (t, $J=4 \mathrm{~Hz}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), $2.08-$ $1.89\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right){ }^{13}{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, DMSO- $d_{6}$ ), $\delta$ ppm: 195.43 ( $2-\mathrm{C}=\mathrm{O}$ ), 153.48, 152.17, $145.19,131.28$ ( $8-\mathrm{C}$ ), 128.92, 128.79 ( $8-\mathrm{CH}$ ), 121.70 ( $2-$ C), $114.60(2-\mathrm{C} \equiv \mathrm{N}), 62.99(2-\mathrm{C}), 42.47\left(2-\mathrm{CH}_{2}-\mathrm{N}\right)$, $36.22(2-\mathrm{CH}), 35.95,26.42,21.29\left(6-\mathrm{CH}_{2}\right)$; analysis: calculated for $\mathrm{C}_{34} \mathrm{H}_{30} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{Cl}_{2}$ (625.54): C, $65.27 ; \mathrm{H}, 4.84$; N, 13.43\%. Found: C, 65.17; H, 4.76; N, 13.51\%.

2-Amino-1-(4-aminophenyl)-4-(4-chlorophenyl)-5-oxo-1,4,5,6, 7,8-hexahydroquinoline-3-carbonitrile (2e). White crystal from ethanol; yield $64 \%, \operatorname{mp} 264^{\circ} \mathrm{C}$.; IR ( $\lambda_{\text {max }}, \mathrm{cm}^{-1}$ ): 3453 $\mathrm{m}, 3334 \mathrm{~m}\left(\mathrm{NH}_{2}\right), 3034 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right)$, 2944-2819 w $\left(\mathrm{CH}_{\text {aliph. }}\right), 2172 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1641 \mathrm{~s}(\mathrm{C}=\mathrm{O}), 1608 \mathrm{~s}(\mathrm{C}=\mathrm{C})$; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 400 MHz, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}$ : 7.37-6.67 (m, 8H, $\mathrm{CH}_{\text {arom. }}$ ), $5.48,5.19\left(\mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{NH}_{2}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right)$, $4.50(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 4.21(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}$ ethanol molecule in the crystal lattice), $3.48-3.45$ ( $\mathrm{q}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ethanol molecule in the crystal lattice), $2.23-2.18\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.04$ $2.00\left(\mathrm{t}, J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{C}=\mathrm{C}\right), 1.86-1.62\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}-\right.$ $\left.\mathrm{CH}_{2}-\mathrm{CH}_{2}\right), 1.09\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right.$ ethanol molecule in the crystal
lattice); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right), \delta \mathrm{ppm}: 195.35$ $(\mathrm{C}=\mathrm{O}), 154.04,152.24,150.27,146.25,131.19$ (5-C), $130.52,129.08,128.72,114.82(8-\mathrm{CH}), 123.69,121.85$ (2C), $112.51(\mathrm{C} \equiv \mathrm{N}), 59.83,56.52$ (2-C), 36.49 (CH), 36.16, 28.15, 21.12, $18.97\left(3-\mathrm{CH}_{2}\right)$; analysis: calculated for $\mathrm{C}_{22} \mathrm{H}_{19} \mathrm{~N}_{4} \mathrm{OCl}$ (390.86): C, 67.59; H, 4.90; N, $14.33 \%$. Found: C, 67.43; H, 4.86; N, 14.37\%.

Synthesis of ethyl-2-[9-(2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5, 6,7,8,9,10-decahydroacridin-10-yl]acetate and ethyl-2-[9-(3-bro mo-2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-de cahydroac ridin-10-yl]acetate (3a,b) (general procedure). To a mixture of 1,3 -cyclohexanedione ( $3.36 \mathrm{~g}, 0.03 \mathrm{~mol}$ ), ethyl glycinate hydrochloride ( $2.09 \mathrm{~g}, 0.015 \mathrm{~mol}$ ), and the appropriate aldehyde; salicaldehyde ( $1.58 \mathrm{~mL}, 0.015 \mathrm{~mol}$ ) or 5-bromo-2-hydroxy benzaldehyde ( $3.015 \mathrm{~g}, 0.015 \mathrm{~mol}$ ) in ethanol $(30 \mathrm{~mL})$, triethylamine $(2.09 \mathrm{~g}, 0.015 \mathrm{~mol})$ was added. The reaction mixture was heated under reflux for 5 h at $80-85^{\circ} \mathrm{C}$ left to cool, and the separated solid was filtered off, dried, and recrystallized from ethanol.

9-(2-hydroxyphenyl)-1,8-dioxo-2,3,4,5,6,7,8,9-octahydroacridi ne-10(1H)-yl)ethylacetate (3a). Yellow crystal from methanol; yield $77 \%$; mp $230-232^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }, \mathrm{cm}^{-1}$ ): 3111 br $(\mathrm{OH}), 3061 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right)$, 2981-2870 w $\left(\mathrm{CH}_{\text {aliph }}\right)$, 1739 vs $\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 1648 \mathrm{~s}, 1630 \mathrm{~s}\left(2 \mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 1593 \mathrm{~s}$ $(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right), \delta \mathrm{ppm}: 9.58$ (s, $1 \mathrm{H}, \mathrm{OH}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.99-6.66\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{\text {arom }}\right)$, 4.99 (s, 1H, CH-Ar), 4.83 (s, 2H, N-CH2-C=O), 4.28$4.22\left(\mathrm{q}, J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 2.90-2.86,2.44-2.40$ ( $\left.\mathrm{t}, J=8 \mathrm{~Hz}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.29-2.27(\mathrm{t}, J=8 \mathrm{~Hz}$, $\left.4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{C}=\mathrm{C}\right), 1.99-1.94,1.82-1.79\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}-\right.$ $\left.\mathrm{CH}_{2}-\mathrm{CH}_{2}\right), 1.28-1.25\left(\mathrm{t}, J=8 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}-$ NMR ( 100 MHz, DMSO- $\left.d_{6}\right), \delta \mathrm{ppm}: 198.00\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right)$, $169.68\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 156.03$ (2-C), 153.47, 132.76 (2-C), $128.51,127.93,120.49,117.22(4-\mathrm{CH}), 114.95$ (2-C), $62.18\left(\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 47.75\left(\mathrm{CH}_{2}-\mathrm{N}\right), 35.98(\mathrm{CH}), 26.07$, 25.71, $20.95\left(6-\mathrm{CH}_{2}\right), 14.40\left(\mathrm{CH}_{3}\right)$; MS ( $\mathrm{m} / \mathrm{z}, \mathrm{I} \%$ ): 395 (95.82) ( $\mathrm{M}^{+}$), 377 (52.37), 339 (25.86), 308 (28.93), 302 (100.00), 290 (74.46), 274 (77.73), 262 (26.18), 199 (62.77); analysis: calculated for $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{NO}_{5}$ (395.44): C, 69.85; H, 6.38; N, 3.54\%. Found: C, 69.91; H, 6.43; N, $3.47 \%$.

9(2-hydroxy-5-bromophenyl)-1,8-dioxo-2,3,4,5,6,7,8,9-octa hydroacridine- $\mathbf{1 0 ( 1 H ) y l}$ )ethyl acetate [18] (3b). Yellow crystal; yield $80 \% ; \mathrm{mp} 242^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }, \mathrm{cm}^{-1}$ ): 3150 ( OH ), $3049 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right), 2980-2888 \mathrm{w}\left(\mathrm{CH}_{\text {aliph. }}\right), 1737$ vs $\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 1631 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 1593 \mathrm{~s}(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 9.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.14-6.65\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{\text {arom. }}\right), 4.96$ (s, $1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}$ ), 4.85 ( $\mathrm{s}, 2 \mathrm{H},-\mathrm{N}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 4.28-4.23 ( $\mathrm{q}, J=8 \mathrm{~Hz}, 2 \mathrm{H},-\mathrm{CH}_{2}-\mathrm{CH}_{3}$ ), 2.89-2.86, 2.46-2.42 (t, $\left.J=8 \mathrm{~Hz}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.31-2.28(\mathrm{t}, J=8 \mathrm{~Hz}$, $\left.4 \mathrm{H}, \quad 2 \mathrm{CH}_{2}-\mathrm{C}=\mathrm{C}\right), \quad 2.00-1.94, \quad 1.84-1.77(\mathrm{~m}, 4 \mathrm{H}, 2$ $\left.\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}\right), 1.28-1.26\left(\mathrm{t}, J=8 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{CH}_{2}-\right.$ $\left.\mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}\right.$, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 197.60$
$\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 169.62\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 156.16(2-\mathrm{C}), 153.55$, $135.48,131.05$ (3-C), $130.66,119.58,114.34$ (3-CH), $111.63(2-\mathrm{C}), 62.02\left(\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 47.77\left(\mathrm{CH}_{2}-\mathrm{N}\right), 36.02$ $(\mathrm{CH}), 26.53,26.12,20.98\left(6-\mathrm{CH}_{2}\right), 14.51\left(\mathrm{CH}_{3}\right) ; \mathrm{MS}$ $(\mathrm{m} / \mathrm{z}, \mathrm{I} \%): 479(0.56), 478(0.86)\left(\mathrm{M}^{+}+2\right), 477(4.68)$, $476(25.75)\left(\mathrm{M}^{+}+1\right), 475(90), 474(30.26)\left(\mathrm{M}^{+}\right), 433$ (1.44), 404 (4.46), 384 (0.9), 300 (1.59), 281 (1.02), 253 (1.56), 206 (4.95), 91 (5.9), 91.95 (3.4), 52 (0.91), 51 (1.42), 50 (0.62); analysis: calculated for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{NO}_{5} \mathrm{Br}$ (474.34): C, 58.23 ; H, 5.11 ; N, 2.95\%. Found: C, 58.15; H, 5.23; N, 2.83\%.

Synthesis of $N^{\prime}$-[3-cyano-9-(4-chlorophenyl)-1-(4-methyl phenyl)-5-oxo-1,4,5,6,7,8-hexa hydroquinolin-2-yl]-N,N-dimethy limidoformamide (4). A solution of compound $\mathbf{4 b}(3.12 \mathrm{~g}$, $0.008 \mathrm{~mol})$ and DMF/DMA ( $1.1 \mathrm{~mL}, 0.008 \mathrm{~mol}$ ) in DMF $(40 \mathrm{~mL})$ was refluxed for 5 h , cooled, and then poured into ice-cold water. The formed precipitate was collected by filtration and recrystallized from ethanol as pale brown crystal yield $93 \% ; \mathrm{mp} 228^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }, \mathrm{cm}^{-1}$ ): 3039 w $\left(\mathrm{CH}_{\text {arom. }}\right), 2970-2807 \mathrm{w}\left(\mathrm{CH}_{\text {aliph }}\right), 2182 \mathrm{~m}(\mathrm{C} \equiv \mathrm{N}), 1635 \mathrm{~s}$ $\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right.$ ketone $), 1610 \mathrm{~s}(\mathrm{C}=\mathrm{N}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 7.67(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}=\mathrm{N}), 7.38-7.13(\mathrm{~m}, 8 \mathrm{H}$, $\mathrm{CH}_{\text {arom. }}$ ), 4.62 (s, $\left.1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}\right), 2.90\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{N}\right), 2.34$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{N}$ ), $2.20\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{Ph}\right.$ ), 2.04-1.99 (t, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), $\quad 1.83-1.80 \quad\left(\mathrm{t}, \quad 2 \mathrm{H}, \quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\right.$ $\mathrm{C}=\mathrm{O}), 1.67-1.63$ (m, 2 $\mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 195.39$ ( $\mathrm{C}=\mathrm{O}$ ), 157.68, $155.64,153.92,145.69,137.99,136.37,131.45,130.11$, $129.64,129.34,128.85,122.06,111.60(\mathrm{C} \equiv \mathrm{N}), 70.71$ (C), $39.96(\mathrm{CH}), 37.97,36.55,34.28\left(3-\mathrm{CH}_{2}\right), 28.31\left(\mathrm{CH}_{3}-\mathrm{Ar}\right)$, 21.27, $21.13 \quad\left(2-\mathrm{CH}_{3}-\mathrm{N}\right)$; analysis: calculated for $\mathrm{C}_{26} \mathrm{H}_{25} \mathrm{~N}_{4} \mathrm{OCl}$ (444.95): C, $70.17 ; \mathrm{H}, 5.67 ; \mathrm{N}, 12.59 \%$. Found: C, 70.21; H; 5.61; N, 12.47\%.

## Synthesis of 2-methyl-9-(4-chlorophenyl)-10-(4-methylphenyl)-

 5,8,9,10-tetrahydropyr-imido $[4,5-b] q u i n o l i n e-4,6(3 H, 7 H)$-dione (5). A solution of compound $\mathbf{2 b}(0.78 \mathrm{~g}, 0.002 \mathrm{~mol})$ in dry acetic anhydride ( 20 mL ) was refluxed for 3 h ; the solvent was evaporated under reduced pressure, and the formed precipitate was collected and recrystallized from DMF as off white crystal, yield $68 \%$; mp $>320^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }$, $\mathrm{cm}^{-1}$ ): $3395 \mathrm{~m}(\mathrm{NH}), 3042 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right), 2951-2859 \mathrm{w}$ $\left(\mathrm{CH}_{\text {aliph. }}\right), 1682 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 1641 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic ketone }}\right)$, $1594 \mathrm{~s}(\mathrm{C}=\mathrm{N})$ ) ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right), \delta \mathrm{ppm}$ : $12.06\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.97(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}$ DMF molecule in the crystal lattice), $7.37-7.17$ (m, 8 H , $\mathrm{CH}_{\text {arom. }}$ ), 5.17 (s, 1H, CH-Ar), 2.9, 2.75 (s, 6H, $2 \mathrm{CH}_{3}$ DMF molecule in the crystal lattice), $2.40\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\right.$ $\mathrm{Ph}), 2.24-2.10\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.03(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}-\mathrm{C}=\mathrm{N}\right), \quad 1.87-1.68 \quad\left(\mathrm{~m}, \quad 4 \mathrm{H}, \quad 2 \mathrm{CH}_{2}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $\left.100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right), \delta \mathrm{ppm}: 195.29\left(\mathrm{C}=\mathrm{O}_{\text {cyclic ketone }}\right)$, 162.74 ( $\mathrm{HC}=\mathrm{O}_{\text {DMF molecule }}$ ), $162.07\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 157.06$, $154.57,153.66,145.55,138.14,136.21,131.00$ (7-C), $130.17,130.05,129.73,128.39(8-\mathrm{CH}), 112.54,101.18$ (2C), $36.21,31.27\left(2-\mathrm{CH}_{3 \mathrm{DMF}}\right.$ molecule $), 36.68(\mathrm{CH}), 32.67$,28.41, $21.60\left(3-\mathrm{CH}_{2}\right), 21.31,21.18\left(2-\mathrm{CH}_{3}\right) ; \mathrm{MS}(\mathrm{m} / \mathrm{z}$, I \%): 433 (1.61) ( $\left.\mathrm{M}^{+}+2\right), 431(5.79)\left(\mathrm{M}^{+}\right), 336(1.24), 320$ (23.81), 279 (1.03), 91.05 (8.46), 65 (6.37); analysis: calculated for $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{Cl}$ (431.91): C, 69.51; H, 5.14; N, $9.73 \%$. Found: C, 69.63 ; H, 5.07 ; N, $9.68 \%$.

Synthesis of 4-(4-chlorophenyl)-1-(4-methylphenyl)-2,5-dioxo-1,2,3,4,5,6,7,8-octahydro-quinolone-3-carboxylic acid (6). A solution of compound $\mathbf{2 b}(2.0 \mathrm{~g}, 0.0051 \mathrm{~mol})$ in concd $\mathrm{H}_{2} \mathrm{SO}_{4}(20 \mathrm{~mL})$ was stirred for 4 h at room temperature, and the reaction mixture was poured into ice-cold water. The formed precipitate was collected, filtered of, washed with water, and recrystallized from ethanol as pale yellow crystal, yield; $89 \% ; \mathrm{mp} 245^{\circ} \mathrm{C}$; IR ( $\left.\lambda_{\text {max }}, \mathrm{cm}^{-1}\right): 3192 \mathrm{br}\left(\mathrm{OH}_{\text {acid }}\right), 3034 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right)$, 2948-2867 w ( $\left.\mathrm{CH}_{\text {aliph. }}\right), 1705 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {acidic }}\right), 1671 \mathrm{~s}$ $\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 1631 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 1601 \mathrm{~s}(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 7.76\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}_{\text {acid }}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.41-7.10\left(\mathrm{~m}, 8 \mathrm{H}, \mathrm{CH}_{\text {arom. }}\right), 4.61$ (d, $J=4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{COOH}), 3.59(\mathrm{~d}, J=4 \mathrm{~Hz}$, $1 \mathrm{H}, \quad \mathrm{CH}-\mathrm{Ar}), \quad 2.37 \quad\left(\mathrm{~s}, \quad 3 \mathrm{H}, \quad \mathrm{CH}_{3}-\mathrm{Ph}\right), \quad 2.30-2.26$ ( $\mathrm{t}, J=8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 2.10-2.05 ( $\mathrm{t}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), $1.92-1.87\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\right.$ $\overline{\mathrm{C}=\mathrm{O}}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}\right.$, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 195.51$ $\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), \quad 169.64 \quad\left(\mathrm{C}=\mathrm{O}_{\text {acid }}\right), \quad 167.62 \quad\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right)$, 156.78, 140.06, 138.53, 135.06, 132.00 (5-C), 130.26, 130.04, 129.31, 129.10 ( $8-\mathrm{CH}$ ), 114.82 (C), 56.04 (C), $37.88(\mathrm{CH}), 36.36,28.08,21.88\left(3-\mathrm{CH}_{2}\right), 21.12\left(\mathrm{CH}_{3}\right)$; MS ( $\mathrm{m} / \mathrm{z}, \mathrm{I} \%$ ): 409 ( 0.27 ) ( $\mathrm{M}^{+}$), 405 ( 0.76 ), 380 ( 0.61 ), 366 (9.14), 364 (29.15), 91 (16.91); analysis: calculated for $\mathrm{C}_{23} \mathrm{H}_{20} \mathrm{NO}_{4} \mathrm{Cl}$ (409.86): C, 67.39 ; $\mathrm{H}, 4.92$; $\mathrm{N}, 3.41 \%$. Found: C, 67.31; H, 4.80; N, 3.48\%.

Synthesis of ethyl [4-(4-chlorophenyl)-1-(4-methylphenyl)-2,5-dioxo-1,2,3,4,5,6,7,8-octa hydroquinoline]-3-carboxylate (7).

Method 1. To a solution of compound 2b ( $2.0 \mathrm{~g}, 5.13$ mmol) in absolute ethanol ( 50 mL ), concd H2SO4 ( 2 mL ) was added then allowed to heat under reflux for 4 h , cooled, and poured into ice-cold water. The formed precipitate was collected, filtered, and recrystallized from ethanol as white crystal, yield $70 \% ; \mathrm{mp} 1600^{\circ} \mathrm{C}$.

Method 2. To a solution of compound $\mathbf{6}(1.58 \mathrm{~g}$, $0.004 \mathrm{~mol})$ in absolute ethanol $(40 \mathrm{~mL})$ was added concd $\mathrm{H}_{2} \mathrm{SO}_{4}(1.5 \mathrm{~mL})$. The reaction mixture was refluxed for 4 h , cooled, and then poured into ice-cold water. The formed precipitate was collected, filtered, and recrystallized from ethanol as white crystal, yield $77 \% ; \mathrm{mp} 160^{\circ} \mathrm{C}$; IR ( $\lambda_{\text {max }}$, $\mathrm{cm}^{-1}$ ): $3252 \mathrm{~m}(\mathrm{NH}), 3032 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right), 2995-2868 \mathrm{w}$ $\left(\mathrm{CH}_{\text {aliph. }}\right), 1739$ vs $\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 1703 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 1647 \mathrm{~s}$ $\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), \quad 1614 \mathrm{~s} \quad(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \quad(400 \mathrm{MHz}$, DMSO- $d_{6}$ ), $\delta$ ppm: $7.40-7.07$ (m, $8 \mathrm{H}, \mathrm{CH}_{\text {arom. }}$ ), 4.68 (d, $J=4 \mathrm{~Hz}, 1 \mathrm{H}, \underline{\mathrm{CH}}-\mathrm{COOEt}), 4.25-4.22(\mathrm{q}, J=4 \mathrm{~Hz}, 2 \mathrm{H}$, $\left.\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 3.92(\mathrm{~d}, J=4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 2.37(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}-\mathrm{Ph}\right), 2.30\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.10-2.06(\mathrm{t}$, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 1.92-1.86 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-$ $\mathrm{C}=\mathrm{O}), \quad 1.25-1.22 \quad\left(\mathrm{t}, \quad 3 \mathrm{H}, \quad-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right) ; \quad{ }^{13} \mathrm{C}-\mathrm{NMR}$
( 100 MHz, DMSO- $d_{6}$ ), $\delta$ ppm: $195.35\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 168.02$ $\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 165.91\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 156.56,138.80,138.47$, 134.51, 132.30 (5-C), 130.43, 129.33, 129.28, 128.77 (8$\mathrm{CH}), 115.36(\mathrm{C}), 62.14\left(\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 55.06,37.04(2-\mathrm{CH})$, 36.28, 27.96, $21.97\left(3-\mathrm{CH}_{2}\right), 21.13\left(\mathrm{CH}_{3} \mathrm{Ar}\right), 14.46\left(\mathrm{CH}_{3}-\right.$ $\mathrm{CH}_{2}$ ); analysis: calculated for $\mathrm{C}_{25} \mathrm{H}_{24} \mathrm{NO}_{4} \mathrm{Cl}$ (437.91): C, 68.56; H, 5.53; N, 3.19\%. Found: C, 68.41; H, 5.65; N, 3.23\%.
Synthesis of ethyl[4-(4-chlorophenyl)-1-(4-methylphenyl)-
(5E)-2-0xo-5-(2-phenyl-hydrazinylidene)-1,2,3,4,5,6,7,8-octahydro
quinoline]-3-carboxylate (8). A mixture of compound 7
$(0.65 \mathrm{~g}, 0.0015 \mathrm{~mol})$, phenyl hydrazine $(0.15 \mathrm{~mL}$,
0.0015 mol ), and catalytic amount of trifloroacetic acid in
ethanol ( 20 mL ) was heated under reflux for 4 h . The
solvent was evaporated under reduced pressure, and the
formed precipitate was collected by filtration and
recrystallized from ethanol as green crystal, yield $76 \%$;
$\mathrm{mp} 138-140^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }, \mathrm{cm}^{-1}$ ): $3252 \mathrm{~m}(\mathrm{NH}), 3099-$
3024 w ( $\mathrm{CH}_{\text {arom. }}$ ), 2981-2836 w ( $\mathrm{CH}_{\text {aliph. }}$ ), 1730 vs
$\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 1666 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 1630 \mathrm{~s}(\mathrm{C}=\mathrm{N}), 1596 \mathrm{~s}$
$(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right)$, $\delta \mathrm{ppm}: 8.97$ (s,
$1 \mathrm{H}, \mathrm{NH}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.44-6.68(\mathrm{~m}, 13 \mathrm{H}$,
$\mathrm{CH}_{\text {arom. }}$ ), 5.06 (d, $J=4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{COOEt}$ ), 4.25-4.20
(q, $\left.J=4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right), 3.83(\mathrm{~d}, J=4 \mathrm{~Hz}, 1 \mathrm{H}$,
$\mathrm{CH}-\mathrm{Ar}), 2.36\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}-\mathrm{Ph}\right), 2.62-2.58(\mathrm{t}, 2 \mathrm{H}$,
$\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 1.89-1.86 ( $\mathrm{t}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-$
$\mathrm{C}=\mathrm{O}), 2.12,1.69\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 1.21-$
$1.19\left(\mathrm{t}, J=4 \mathrm{~Hz}, 3 \mathrm{H},-\mathrm{CH}_{2}-\mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$,
DMSO- $\left.d_{6}\right), \quad \delta \quad$ ppm:168.67 $\quad\left(\mathrm{C}=\mathrm{O}_{\text {ester }}\right), 165.27$
$\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 146.60,142.01,139.56,139.45,137.94$,
135.15, 131.93, 115.87 (8-C), 130.14, 129.96, 129.53,
129.21, 128.68, 118.91, $112.86(13-\mathrm{CH}), 61.72\left(\mathrm{CH}_{2}{ }^{-}\right.$
$\left.\mathrm{CH}_{3}\right), 55.56,38.54(2-\mathrm{CH}), 26.78,23.35,21.30\left(3-\mathrm{CH}_{2}\right)$,
$21.12\left(\mathrm{CH}_{3} \mathrm{Ar}\right), 14.52\left(\mathrm{CH}_{3}-\mathrm{CH}_{2}\right)$; analysis: calculated
for $\mathrm{C}_{31} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Cl}(528.04)$ : $\mathrm{C}, 70.50 ; \mathrm{H}, 5.73 ; \mathrm{N}$,
$7.95 \%$. Found: C, $70.46 ; \mathrm{H}, 5.78 ; \mathrm{N}, 7.89 \%$.

Synthesis of [9-(3-bromophenyl)-2,3,4,5,6,7,8,9-octahydro acridin- $10(1 \mathrm{H})$-yl acetic acid (9). To a solution of compound 3b $(2.37 \mathrm{gm}, 0.005 \mathrm{~mol})$ in ethanol $(40 \mathrm{~mL})$ was added NaOH solution $(0.4 \mathrm{gm}, 0.01 \mathrm{~mol}$ in 5 mL water). The reaction mixture was heated under reflux for 5 h , cooled, poured to cold water, and acidified with concd HCl . The separated solid was filtered off, dried, and recrystallized from ethanol, yield $97 \%$; mp 237-239 C ; IR $\left(\lambda_{\max }, \mathrm{cm}^{-1}\right)$ : $3444 \mathrm{br}\left(\mathrm{OH}_{\text {acid }}\right), 3110 \mathrm{br}$ $\left(\mathrm{OH}_{\text {phenolic }}\right), 3015 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right), 2972-2882 \mathrm{w}\left(\mathrm{CH}_{\text {aliph. }}\right)$, $1707 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {acid }}\right), 1625 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 1585 \mathrm{~s}(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 13.29$ (br, 1 H , $\mathrm{OH}_{\text {acidic }}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 9.77\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}_{\text {phenolic }}\right.$, disappeared by $\mathrm{D}_{2} \mathrm{O}$ ), $7.10-6.67\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{\text {arom. }}.\right), 4.98$ (s, 1H, CH-Ar), 4.71 (s, 1H, N-CH2-C=O), 4.47 (q, 2H, $\mathrm{CH}_{2}$ ethanol molecule in the crystal lattice), $3.2(\mathrm{~s}, 1 \mathrm{H}$, OH disappeared by $\mathrm{D}_{2} \mathrm{O}$, ethanol molecule in the crystal lattice), $2.9,2.5\left(\mathrm{t}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \underline{\mathrm{CH}}_{2}-\mathrm{C}=\mathrm{O}\right), 2.30$
( $\mathrm{t}, \quad 4 \mathrm{H}, \quad 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), $1.98, \quad 1.84 \quad(\mathrm{~m}, 4 \mathrm{H}$, $\left.2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 1.08\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right.$ ethanol molecule in the crystal lattice); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}\right.$, DMSO- $d_{6}$ ), $\delta \quad$ ppm: $197.46 \quad\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 170.98 \quad\left(\mathrm{C}=\mathrm{O}_{\text {acid }}\right)$, $156.15,153.56,135.66$ (6-C), $131.19,130.56,119.50$ (3$\mathrm{CH}), \quad 114.40, \quad 111.67 \quad(4-\mathrm{C}), \quad 56.52 \quad\left(\mathrm{CH}_{2} \quad\right.$ ethanol molecule), $47.83\left(\mathrm{CH}_{2}-\mathrm{N}\right), 36.08(\mathrm{CH}), 26.55,26.22$, $21.05\left(6-\mathrm{CH}_{2}\right), 19.01\left(\mathrm{CH}_{3}\right.$ ethanol molecule); analysis: calculated for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{NO}_{5} \mathrm{Br}$ (446.29): C, $56.51 ; \mathrm{H}, 4.52$; N, $3.13 \%$. Found: C, 56.38 ; H, 4.43; N, 3.27\%.

Synthesis of 2-[9-(2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5, $6,7,8,9,10$ decahydroacridin-10-yl]acetohydrazide and 2-[9-(3-bromo-2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-decahy dro-acridin-10-yl]acetohydrazide (10a,b) (general procedure).
A solution of compound $\mathbf{3 a}, \mathbf{b}(0.005 \mathrm{~mol})$ in hydrazine hydrate ( 20 mL ) was refluxed for 2 h , cooled, and then triturated with ethanol. The precipitated solid obtained was filtered off, dried, and recrystallized from DMF to furnish compound 10a,b.

2-[9-(2-Hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-deca hydroacridin-10-yl]acetohyd-razide (10a). Yellow powder from DMF; yield $90 \%$; decomposed at $264^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }$, $\left.\mathrm{cm}^{-1}\right): 3362 \mathrm{br}(\mathrm{OH}), 3339 \mathrm{~m}, 3251 \mathrm{~m}\left(\mathrm{NH}_{2}, \mathrm{NH}\right), 3051$ w ( $\mathrm{CH}_{\text {arom. }}$ ), 2926-2867 w $\left(\mathrm{CH}_{\text {aliph. }}\right), 1668 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {amide }}\right)$, $1627 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right) 1582 \mathrm{~s}(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 11.05$ (br, $1 \mathrm{H}, \mathrm{NH}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 9.07\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.98-6.56$ ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{CH}_{\text {arom. }}$ ), 5.63 (broad, $2 \mathrm{H}, \mathrm{NH}_{2}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 5.20(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 4.22\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{N}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right)$, 2.51, 2.38-2.34 ( $\mathrm{t}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 2.31-2.29, 2.07-2.01 ( $\left.\mathrm{t}, \quad 4 \mathrm{H}, \quad 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), \quad 1.82-1.78$, 1.63-1.60 (m, 4H, $\quad 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (100 MHz, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 195.74\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right)$, $169.35\left(\mathrm{C}=\mathrm{O}_{\text {amidic }}\right), 154.85,148.49,140.22,134.62$ (4-C), $128.55,126.72,119.29,116.95(4-\mathrm{CH}), 110.92$ (2-C), $46.77\left(\mathrm{CH}_{2}-\mathrm{N}\right), 27.93(\mathrm{CH}), 25.23,22.47,20.82$ (6-CH $)_{2}$; MS ( $\mathrm{m} / \mathrm{z}, \mathrm{I} \%$ ): $381(0.18)\left(\mathrm{M}^{+}\right), 269(0.96), 199$ (1.07), 180 (2.13), 167 (1.06), 94 (100.00), 66 (33.85); analysis: calculated for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{4}$ (381.42): $\mathrm{C}, 66.12$; H, 6.09; N, 11.01\%. Found: C, 66.02; H, 6.23; N, 11.18\%.

2-[9-(3-Bromo-2-hydroxyphenyl)-1,8-dioxo-1,2,3,4,5,6,7,8, 9,10decahydroacridin-10-yl] acetohydrazide (10b). Yellow powder from DMF; yield $92 \%$; decomposed at $254^{\circ} \mathrm{C}$; IR $\left(\lambda_{\max }, \mathrm{cm}^{-1}\right): 3534$ br $(\mathrm{OH}), 3333 \mathrm{~m}, 3209 \mathrm{~m}\left(\mathrm{NH}_{2}\right.$, NH), 3087-3046 w ( $\left.\mathrm{CH}_{\text {arom. }}\right)$, 2951-2870 w ( $\mathrm{CH}_{\text {aliph. }}$ ), $1679 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {amide }}\right), 1637 \mathrm{~s}\left(\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right), 1617 \mathrm{~s}(\mathrm{C}=\mathrm{C})$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right)$, $\delta \mathrm{ppm}: 11.65(\mathrm{br}, 1 \mathrm{H}$, NH , disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 9.13(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.01-6.50\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{\text {arom. }}\right)$, 5.65 (br, 2 H , $\mathrm{NH}_{2}$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 5.14(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-\mathrm{Ar}), 4.25$ ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 2.58-2.54, 2.40-2.36 (t, 4 H , $\left.2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), \quad 2.29-2.21, \quad 2.07-2.02 \quad(\mathrm{t}, \quad 4 \mathrm{H}$, $\left.2 \mathrm{CH}_{2} \mathrm{CH}_{2} \overline{\mathrm{CH}}_{2}-\mathrm{C}=\mathrm{O}\right), \quad 1.82-1.79, \quad 1.62-1.60 \quad(\mathrm{~m}, 4 \mathrm{H}$, $2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right)$,
$\delta$ ppm: $194.86\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right) 168.97\left(\mathrm{C}=\mathrm{O}_{\text {amide }}\right)$, 154.49, 149.10, 140.86, 137.41 (4-C), $130.96,130.03,129.55$ $(3-\mathrm{CH}), 119.26,110.58(2-\mathrm{C}), 46.72\left(\mathrm{CH}_{2}-\mathrm{N}\right), 28.56$ $(\mathrm{CH}), 25.33,22.52,20.82\left(6-\mathrm{CH}_{2}\right)$; analysis: calculated for $\mathrm{C}_{21} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{Br}$ (460.32): $\mathrm{C}, 54.79 ; \mathrm{H}, 4.82 ; \mathrm{N}$, $9.13 \%$. Found: C, 54.64 ; H, 4.75 ; N, $9.27 \%$.

Synthesis of 1 -\{ $\boldsymbol{N}^{\prime}$-(4-chlorobenzylidine)[9-(2-hydroxyphe nyl)-1,8-diox $0-2,3,4,5,6,7,8,9$-octahydroacridin-10(1H)-yl]\}ace tohydrazide (11). A mixture of compound 10a (0.38 g, 0.001 mol ), p-chlorobenzaldehyde ( $0.14 \mathrm{~g}, 0.001 \mathrm{~mol}$ ), and DMF $(4 \mathrm{~mL})$ in dioxane $(20 \mathrm{~mL})$ were heated under reflux for 5 h . The solvent was evaporated under reduced pressure, and the formed precipitate was filtered off and recrystallized from ethanol as brown crystal in yield $50 \%$; $\mathrm{mp} 288-290^{\circ} \mathrm{C}$; IR ( $\lambda_{\max }, \mathrm{cm}^{-1}$ ): $3351 \mathrm{br}(\mathrm{OH}), 3283 \mathrm{~m}$ (NH), $3050 \mathrm{w}\left(\mathrm{CH}_{\text {arom. }}\right), 2932-2864 \mathrm{w}\left(\mathrm{CH}_{\text {aliph. }}\right), 1622 \mathrm{~s}$ ( $\mathrm{C}=\mathrm{O}_{\text {amide }}$ ) and ( $\mathrm{C}=\mathrm{O}_{\text {cyclic }}$ ), $1561 \mathrm{~s}(\mathrm{C}=\mathrm{N}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ), $\delta \mathrm{ppm}: 10.47(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}$, disappeared by $\mathrm{D}_{2} \mathrm{O}$ ), $8.47(\mathrm{~s}, 1 \mathrm{H}, \mathrm{N}=\mathrm{CH}-\mathrm{Ph}), 7.87-$ $7.85,7.53-7.51\left(\mathrm{~d}, 4 \mathrm{H}, \mathrm{CH}_{\text {arom. }}\right), 6.97-6.65(\mathrm{~m}, 4 \mathrm{H}$, $\left.\mathrm{CH}_{\text {arom. }}\right), 5.34\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 5.29(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}-$ $\mathrm{Ar}), 4.3\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$, an ethanol molecule in the crystal lattice), $3.46\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}\right.$, disappeared by $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.2(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{OH}$ an ethanol molecule in the crystal lattice), 3.173.04, 2.70-2.56 (t, 4H, $2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}$ ), 2.33-2.29, 2.26-2.23 (t, 4H, 2 $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 2.03-1.93,1.86-$ $1.59\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}-\mathrm{C}=\mathrm{O}\right), 1.06\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}\right.$ an ethanol molecule in the crystal lattice); ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (100 MHz, DMSO- $d_{6}$ ), $\delta \mathrm{ppm}: 195.94\left(2-\mathrm{C}=\mathrm{O}_{\text {cyclic }}\right)$, $168.68\left(\mathrm{C}=\mathrm{O}_{\text {amide }}\right), 157.96(\mathrm{C}), 155.71(\mathrm{CH}=\mathrm{N}), 154.02$, 151.54, 135.71, 134.02, 133.88 (5-C), 130.05, 129.42, $128.28,127.63,120.04,117.32$ ( $8-\mathrm{CH}$ ), 111.64, 110.56, $56.50(3-\mathrm{C}), 36.46\left(\mathrm{CH}_{2}-\mathrm{N}\right), 26.88(\mathrm{CH}), 25.89,25.80$, 25.49, 21.01, 20.40, $19.01\left(6-\mathrm{CH}_{2}\right)$; analysis: calculated for $\mathrm{C}_{28} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{Cl}$ (503.97): $\mathrm{C}, 66.72 ; \mathrm{H}, 5.21 ; \mathrm{N}$, $8.33 \%$. Found: C, $66.84 ;$ H, 5.17 ; N, $8.46 \%$.

## REFERENCES AND NOTES

[1] Beagley, P.; Blackie, M. A. L.; Chibale, K.; Clarkson, C.; Meijboom, R.; Moss, J. R.; Smith, P.; Su, H. Dalton Trans 2003, 2003, 3046.
[2] Mohammed, S. T. M.; Dina, A. B.; Reda, M. A. Int J Org Chem 2012, 2, 49.
[3] Sawada, Y.; Kayakiri, H.; Abe, Y.; Mizutani, T.; Inamura, N.; Asano, M.; Hatori, C.; Aramori, I.; Oku, T.; Tanaka, H. J Med Chem 2004, 47, 2853.
[4] Ma, Z.; Hano, Y.; Nomura, T.; Chen, Y. Bioorg Med Chem Lett 2004, 14, 1193.
[5] Denton, T. T.; Zhang, X.; Cashman, J. R. J Med Chem 2005, 48, 224.
[6] Fokialakis, N.; Magiatis, P.; Chinou, L.; Mitaku, S.; Tillequin, F. Chem Pharm Bull 2002, 50, 413.
[7] Puchakayala, S.; Kalyana, S.; Shalini, D.; Annamalai, S. K. Electrochim Acta 2014, 147, 62.
[8] Hidayat, H.; Sabine, S.; Salem, R. S.; Achim, H.; Karsten, K. Eur J M Chem 2012, 54, 936.
[9] Fossa, P.; Mosti, L.; Menozzi, G.; Marzano, C.; Baccichetti, F.; Bordin, F. Bioorg Med Chem 2002, 10, 743.
[10] Ryckebusch, A.; Derprez-Poulain, R.; Maes, L.; DebreuFontaine, M. A.; Mouray, E.; Grellier, P.; Sergheraert, C. J Med Chem 2003, 46, 542.
[11] Morgan, L. R.; Jursic, B. S.; Hooper, C. L.; Neumann, D. M.; Thangaraj, K.; Leblanc, B. Bioorg Med Chem Lett 2002, 12, 3407.
[12] Ho, Y. L.; Joachim, H.; Amanda, C.; David, H.; Adam, N.; Stephen, P. M. Chem Commun 2014, 50, 10222.
[13] Nnamonu, L. A.; Agwada, V. C.; Nwadinigwe, C. A. Int J Org Chem 2013, 3, 229.
[14] Slusher, S.; Rissolo, K. C.; Jackson, P. F.; Pullan, L. M. J Neural Transm 1994, 97(3), 175.
[15] Yoneda, Y.; Suzuki, T.; Ogita, K.; Han, D. J Neurochem 1994, 60(2), 634.
[16] Nishiyama, T.; Hashiguchi, Y.; Sakata, S.; Sakaguchi, T. Polym Degrad Stab 2003, 79(2), 225.
[17] Mohamed, S. K.; Akkurt, M.; Jasinski, J. P.; Abdallah, O. A.; Albayati, M. R. Acta Crystallogr 2015a, E71, o949.
[18] Sangani, C. B.; Makawana, J. A.; Zhang, X.; Teraiya, S. B.; Lin, L.; Zhu, H. L. Eur J Med Chem 2014, 76, 549.
[19] Martins, M. A. P.; Rossatto, M.; Prola, L. D. T.; Pizzuti, L.; Moreira, D. N.; Campos, P. T.; Frizzo, C. P.; Zanatta, N.; Bonacorso, H. G. Ultrason Sonochem 2012, 19, 227.
[20] Mohamed, S. K.; Akkurt, M.; Jasinski, J. P.; Abdelhamid, A. A.; Tamam, A. H.; Albayati, M. R. Acta Crystallogr 2015b, E71, 0989.

